The effect of dentin pre-treatment with activated riboflavin on the bond strength of a two-step self-etch adhesive system

Abstract

Background. The cross-linking of collagen fibers in the hybrid layer has been suggested as a way to create more durable bonds.

Objectives. This study evaluated the effect of visible light-activated riboflavin (RF) as a cross-linking agent on the durability of the dentin microtensile bond strength (μTBS) in a 2-step self-etch (SE) adhesive system.

Material and methods. The occlusal surfaces of 21 human premolar teeth were ground down to expose the dentin, and were then randomly divided into 3 groups. The Clearfil® SE Bond was used in the control group. In the RF/BL group, a 0.1 wt% aqueous solution of RF was applied to the dentin surface before applying the adhesive and irradiating with blue light (BL) for 2 min. In the 3rd group, the RF-P/BL group, the RF powder was added to the adhesive primer (P) at a concentration of 0.1 wt%. The teeth were built up using composite resin. After thermocycling, 14 resin–dentin beams from each group were prepared and stored in water for 3 months. The μTBS was determined and the data was analyzed using a linear model with a generalized estimating equation (GEE) ($p = 0.05$).

Results. The highest and the lowest μTBS belonged to the control group (41.15 ±3.50 MPa) and the RF-P/BL group (19.84 ±3.80 MPa), respectively. The mean μTBS in the control group was significantly higher than in the RF/BL and RF-P/BL groups ($p < 0.001$), but no significant difference was found between the RF/BL and RF-P/BL groups ($p = 0.598$).

Conclusions. Pre-treatment of dentin surfaces with RF activated with BL had a negative impact on the μTBS of the Clearfil SE Bond as a 2-step SE adhesive.

Key words: collagen, riboflavin, cross-linking agents, dentin bonding agents, photoactivation

Słowa kluczowe: kolagen, ryboflawina, środki siecujące, zębinowe środki wiążące, fotoaktywacja
Introduction

The disintegration of an adhesive interface over time is an important factor influencing the longevity of adhesive restorations.\(^1\) Unstable polymers within the hybrid layer may cause collagen fibers to be denuded. These fibers are highly susceptible to hydrolytic and mechanical fatigue as well as degradation via the collagenolytic activity of host-derived proteases.\(^2\)

Etch and rinse (E&R)\(^3\) and self-etch (SE)\(^4\) adhesives activate dentin matrix metalloproteinases (MMPs) and may be responsible – at least in part – for the gelatinolytic activity observed in the hybrid layer.\(^5,6\) Chlorhexidine has been suggested as a strong MMP inhibitor which has antibacterial properties and, when incorporated into SE adhesives, increases the long-term durability of the adhesive without jeopardizing its mechanical properties and bond strength.\(^7,8\) Matrix metalloproteinases do not undergo turnover in dentin. Thus, inactivating these organic enzymes through the use of cross-linking agents may increase durability and may be even more effective than using MMP inhibitors like chlorhexidine.\(^9,10\)

Increasing the cross-linking of collagen can be done by chemical or physical (photo-oxidative) means. The chemical approach includes the use of different types of cross-linking solutions, while the physical (photo-oxidative) approach is comprised of light exposure – particularly ultraviolet radiation.\(^9,11\) The covalent cross-links created by the external cross-linkers are very stable and deactivate the active sites of dentin proteases by decreasing the mobility of the active sites or by converting the negatively-charged ionized carboxyl groups into positively-charged amides.\(^12\) Moreover, the cross-linking technique has been suggested as a way to improve collagen's stability and resistance to degradation within the demineralized dentin matrix.\(^13–15\) Increasing the cross-linking of collagen fibrils before applying an adhesive agent may increase bonding durability.\(^2,16\) Despite the promising results of the application of cross-linkers, there are still some problems which need to be solved before they can be extensively used in clinical situations. For instance, glutaraldehyde is effective\(^13\) as a collagen cross-linker, but is also toxic.\(^12\) Grape seed extract is also effective\(^10\) and may increase the immediate bond strength to dentin.\(^17\) However, it causes a brown discoloration in dentin and the long-term durability of the resultant bond has yet to be determined.\(^12\)

Riboflavin (vitamin B2; RF) is a strong free-radical producer under light activation with peak absorbance at 270 nm, 366 nm and 445 nm.\(^18\) It significantly increases the cross-linking of type I collagen.\(^19\) When RF is photoactivated with ultraviolet A light (UVA), it produces free radicals and reactive oxygen species (ROS) such as superoxide anion (\(O_2^-\)) and hydroxyl radical (\(' OH\)).\(^20\)

Some studies have shown that in E&R adhesive systems, cross-linking dentin collagen with the use of RF/UVA increases the immediate bond strength, decreases nanole-akage,\(^9,16,21\) stabilizes the adhesive interface, and inhibits the activity of dentin MMPs.\(^9,22\) Furthermore, using RF to change the dentin collagen matrix stabilizes the fibrillar collagen network, promotes resin infiltration and hybrid layer formation\(^23\) and improves the mechanical properties and mechanical stability of a demineralized dentin substrate against hydrolytic or collagenolytic degradation.\(^21,23–25\) It has been reported that visible blue light-activated riboflavin (RF/BL) enhances the resistance to biodegradation and improves the mechanical properties of dentin collagen, bond strength, durability, and interface integrity after short-term water storage.\(^26\) Based on the literature, some protocols of RF application have succeeded in increasing microtensile bond strength (μTBS), while others demonstrated bond strength similar to or lower than that of the control group.\(^21,23,25,26\)

Some previous studies have demonstrated that the acidic resin components present in E&R\(^3\) and SE\(^4\) adhesives increase the collagenolytic and gelatinolytic activity of a completely or partially demineralized collagen matrix. Moreover, it has been reported that SE adhesives upregulate MMP-2 synthesis by human odontoblasts\(^27\) and mild acidic resin monomers can activate MMPs by inhibiting the tissue inhibitor of metalloproteinase-1 in TIMP–MMP complexes.\(^28\) Alternatively, these acidic resin monomers may activate latent forms of MMPs (proMMPs) via a cysteine-switch mechanism.\(^29\)

When choosing a cross-linker or a photoactivation method, some clinical and safety aspects should be considered. The blue light (BL) produced by the highly popular conventional halogen–tungsten curing units (QTH) can be a good alternative to UVA light.\(^26\) Also, the need for a separate curing device (UVA) or for a separate curing process of the cross-linker using BL for optimal results\(^9,26\) demands further investigation in order to achieve a clinically acceptable and applicable technique.

This study sought to assess the in vitro effect of visible light-activated RF on the μTBS of a two-step SE dentin adhesive after thermocycling and 3 months of water storage. The null hypothesis was that visible light-activated RF – whether separately or incorporated into the adhesive – would have no effect on the μTBS of a two-step SE adhesive system after thermocycling and 3 months of water storage.

Material and methods

Sample preparation

This experimental study was conducted on 21 sound human premolar teeth extracted in orthodontic treatment. We obtained verbal consent from the patients to use their extracted teeth in this in vitro study. The research protocol was approved by the institutional ethics committee of the Vice-Chancellor of Research, Hamadan...
University of Medical Sciences, Iran (protocol No. 2015-16p 372). This research was conducted in full accordance with the Declaration of Helsinki.

The teeth were stored in a 0.2% thymol solution until 1 week prior to the start of testing, when the teeth were stored in distilled water. The occlusal enamel of the teeth was removed using a trimmer apparatus with water cooling to expose an area of dentin approx. 5 mm in diameter. The surfaces of the specimens were polished with moist 600-grit silicon carbide papers (Matador® 991A Soflex; Starcke GmbH & Co., Melle, Germany) and the specimens were randomly divided into 3 groups of 7 in each group.

In the control group, Clearfil® SE Bond (Kuraray Medical Inc., Okayama, Japan) was applied as a two-step SE adhesive system to the prepared dentin surfaces, according to the manufacturer’s instructions.

In the RF/BL group, based on previous studies, a 0.1 wt% RF solution was prepared by dissolving 100 mg of riboflavin-5-phosphate powder (Sigma-Aldrich, St. Louis, USA) into 100 mL of distilled water and then adjusting the pH of the RF solution to pH = 3.19,28 The prepared 0.1 wt% RF solutions were stored in lightproof test tubes at room temperature (24°C) in order to prevent light activation. Using a sampler, 15 μL of the RF solution was applied to the dentin surfaces. The treated dentin surfaces were gently air-dried and photoactivated using the BL generated by QTH light curing unit (Optilux® 501; Demetron Kerr, Danbury, USA) with an output light spectrum of 375–520 nm, a minimum light intensity of 600 mW/cm² and a light guide diameter of 8 mm for 2 min. The tip of the light guide was placed as close to the surface as possible. Clearfil SE primer and bonding agent were then applied to the prepared dentin surface and cured according to the manufacturer’s instructions.

In the RF-P/BL group, a riboflavin–primer mixture (RF-P) was created by dissolving RF powder in the primer until reaching a concentration of 0.1 wt%. The dentin surface was treated with the RF-P for 20 s, air-dried for 5 s and photoactivated with BL for 2 min. Then, Clearfil SE Bond adhesive was applied and cured for 10 s.

All of the teeth were built up by using 4 mm of composite resin (Filtek® Z250; 3M ESPE, St. Paul, USA) in 1-millimeter increments. After being stored for 24 h in distilled water at 37°C, the prepared teeth were subjected to 5,000 thermal cycles (5–55°C, 15 s dwell time). To assess the μTBS, the teeth were then sectioned using a low-speed diamond saw with water cooling in order to obtain resin–dentin beams measuring 1 × 1 mm. Two resin–dentin beams were selected from the middle part of each specimen. Thus, each group had 14 specimens. The beams were then subjected to 3 months of water storage at 37°C.

**Bond strength evaluation**

Using cyanoacrylate glue, the beams were mounted on a custom-made metal jig in a universal testing machine ( STM-20; Santam, Tehran, Iran). The μTBS was measured with a 50 N load cell at a crosshead speed of 1 mm/min.

**Failure mode evaluation**

All the fractured samples were evaluated twice under a stereomicroscope (SZ40; Olympus, Tokyo, Japan) at ×40 magnification to determine the mode of failure (adhesive, cohesive in dentin, cohesive in composite resin, and mixed) by 1 operator.

**Sodium dodecyl sulfate – polyacrylamide gel electrophoresis analysis**

The cross-linking effect of the prepared RF solutions on type I collagen was evaluated with sodium dodecyl sulfate polyacrylamide gel electrophoresis (SDS-PAGE). Collagen mixed with distilled water served as the negative control group and 0.1% RF/UVA (pH = 7) was used as the positive control. The samples were diluted with the sample buffer and subjected to SDS-PAGE (8%) at 150 V; the gel was stained with Coomassie blue and then photographed.

**Micromorphological assessment of the resin–dentin interface using scanning electron microscopy**

Two extra resin–dentin beams from each group were selected and prepared for micromorphological assessment of the resin–dentin interface using scanning electron microscope (SEM). These beams were mounted into acrylic resin. The mounted beams were polished with 600-, 800-, 1200-, 1500- and 3500-grit silicon carbide papers (Matador 991A Soflex; Starcke GmbH & Co KG). The specimens were then etched with 35% phosphoric acid for 20 s, rinsed for 20 s and dried with oil-free compressed air. The beams were immersed in a 5.25% sodium hypochlorite solution for 20 min and were thoroughly rinsed under running water for 5 min. The specimens were then dehydrated using increasing concentrations of ethanol – i.e., 33%, 50%, 70%, and 85% – for 15 min each, followed by 90%, 95% and 100% ethanol for 10 min each. After drying, the beams were mounted and sputter-coated with a gold–palladium alloy for 180 s (Sputter Coater SC 7620; Quorum Tech, Lewes, UK). The specimens were then evaluated under SEM (LEO 1450VP; Carl Zeiss, Oberkochen, Germany) at ×5,000 magnification.

**Statistical analysis**

The results of μTBS testing were analyzed with one-way analysis of variance (ANOVA) for the “dentin pretreatment” factor (without RF, with RF/BL or with RF-P/BL), followed by Tukey’s post-hoc pairwise comparison tests. Statistical significance was set at p < 0.05.
**Results**

In all columns, bands were seen at about 130 KDa, belonging to α1 and α2 monomers; in all groups, apart from the negative control group, well residuals were also seen on top of the stacking gel (Fig. 1).

The highest and the lowest bond strength belonged to the control group (41.15 ±3.50) and the RF-P/BL group (19.84 ±3.80), respectively. The generalized estimating equation (GEE) model revealed that the mean μTBS in the control group was significantly higher than in the RF/BL and RF-P/BL groups (p < 0.001); however, the difference in the mean μTBS between the RF/BL and the RF-P/BL groups was not statistically significant (p = 0.598). The mean bond strength values and the frequency distribution of the mode of failure in the study groups are shown in Table 1.

Figure 2 shows the SEM images of the resin–dentin interface in selected samples from different groups after 5,000 thermal cycles and 3 months of water storage. A uniform hybrid layer and long resin tags with relatively regular distribution and straight orientation were clearly seen in the control group (without RF). However, in the RF/BL group, a hybrid layer was formed in only a few sites and there were only a few very fine resin tags with irregular distribution. In the RF-P/BL group, the hybrid layer was less uniform. The resin tags varied in height and showed a more irregular orientation compared to the control group (Fig. 2).

**Discussion**

In SDS-PAGE analysis, both the RF/BL and RF-P/BL groups showed stronger α-bonds and fewer fragments than the control group, which indicates the formation of intermolecular cross-links (Fig. 1). There were insoluble residuals on the wells in both groups that were identified as giant cross-linked collagen molecules. This finding was in line with the results of Chiang et al.21

The μTBS measured in the group without RF in our study (41.15 MPa) was similar to the value reported in previous studies using a similar test30,31 and the failure mode was mostly cohesive in the composite resin. The μTBS value in the presence of RF alone (p < 0.001) or in combination with the primer (p < 0.001) was lower than that in the control.

**Table 1.** Mean value (SD) of bond strength (in MPa) and failure modes of μTBS test

<table>
<thead>
<tr>
<th>Group</th>
<th>Surface pretreatment</th>
<th>Mean of microtensile bond strength</th>
<th>Failure mode (A/M/CR/CD)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control</td>
<td>no pretreatment</td>
<td>41.15 (3.50)a</td>
<td>0/5/7/2</td>
</tr>
<tr>
<td>RF/BL</td>
<td>application and photoactivation of RF% 0.1 (pH = 3)</td>
<td>20.79 (4.41)b</td>
<td>2/12/0/0</td>
</tr>
<tr>
<td>RF-P/BL</td>
<td>application and photoactivation of riboflavin–primer mixture</td>
<td>19.84 (3.80)b</td>
<td>2/12/0/0</td>
</tr>
</tbody>
</table>

SD – standard deviation; RF – riboflavin; RF-P – riboflavin–primer mixture; BL – blue light; A – adhesive; M – mixed; CR – cohesive in resin composite; CD – cohesive in dentin. Values with different uppercase letters indicate significant differences according to Tukey’s test (p < 0.05).
group; therefore, the null hypothesis of the study was upheld. No significant difference was found in the μTBS reduction in the understudy specimens between the 2 modes of application for the cross-linkers (p = 0.598). The majority of failures observed in these 2 groups were mixed.

One advantage of inactivating proteolytic enzymes in a dentin matrix by a cross-linking agent is its non-specific mechanism, since it cross-links all the MMPs present in dentin and collagen fibers. These cross-links are composed of covalent bonds stabilized over time. Riboflavin is a non-toxic vitamin used as a food coloring agent. Moreover, its light yellow color makes it more suitable for dental purposes than other photosensitizing chromophores, such as porphyrins, rose bengal and methylene blue.

In contrast to our findings, some studies have reported that dentin surface pretreatment with UVA-activated RF increased μTBS immediately and after storage in comparison with a control group. The difference between our results and those of previous studies may be attributed to the duration of exposure and the source of light used for activation, as well as the storage conditions.

As with our study, some previous investigations have reported the inefficacy of RF or even its adverse effects on bond strength. Fawzy et al. reported that using 1% RF and single-step photoactivation with BL simultaneously with the bonding produced lower μTBS after 24 h and after 4 months of water storage compared to the control group. Cova et al. reported that when comparing immediate and delayed bond strength, all specimens cross-linked with RF still demonstrated some level of reduction in the mean μTBS after 12 months. It should be noted that all of these studies were conducted on E&R adhesives and did not evaluate SE adhesives.

A reduction in bond strength when RF is used may be due to the strengthening effect of hydration by RF/UVA. The increased hydration of activated RF may be unfavorable to the formation of a stable hybrid layer. Riboflavin can compromise the function of the adhesive system by increasing water sorption from dentin, diluting the primer and maintaining the water content of the primer.

A residual weak collagen matrix within the hybrid layer may be another possible reason for the reduced μTBS in specimens treated with RF after thermocycling and water storage found in our study. As with non-specific inhibitors, the main drawback of cross-linkers used to inactivate MMPs is that they leave a water-rich, resin-deficient collagen matrix with poor mechanical properties within the hybrid layer. These very fragile collagen fibrils are susceptible to creep and subsequent fatigue rupture after long-term functioning. It should be noted that in this study, in contrast to the E&R systems used in previous studies, a SE adhesive system was used. In the process of SE adhesive system application, the smear layer containing proteins, collagen and debris is not eliminated from the dentin surface by the application of acidic monomers. Thus, applying activated RF can cross-link the residual proteins and collagens in the smear layer and cover the dentin surface during the application of acidic monomers of the SE adhesive, obstructing the path of the resin, hindering its penetration into the spaces between collagen fibers. The observation of a partially or deficiently formed hybrid layer and sparse fine resin tags on the SEM imagery from the RF/BL group can confirm this theory. Collagen fibrils which are not fully covered by the resin monomers during the bonding process are susceptible to mechanical fatigue as well as hydrolytic and enzymatic degradation by collagenolytic enzymes.

In the RF-P/BL group, a hybrid layer was formed and the length and distribution of resin tags was less uniform than in the control group. However, due to the simultaneous use of RF and primer, the inhibition of resin penetration was not observed in the respective SEM imagery. In this group, RF was incorporated into the Clearfil SE Bond primer until a 0.1% concentration was reached, and after RF-P was applied to the dentin surface, light activation was carried out for 2 min. It has been reported that treating dentin powder with Clearfil SE Bond primer for 2 min instead of 20 s not only increases collagenolytic activity, but also disables enzymatic inhibition to the point where only 2% chlorhexidine (CHX) can still demonstrate a significant inhibitory effect. Therefore, the reduction in mean μTBS observed in RF-P/BL group may be due to the long duration of dentin treatment with Clearfil SE Bond primer before the application of the bonding agent or due to the low concentration of RF used. Further studies are required to assess the surface micromorphology of treated dentin and to find the best concentration of RF in the SE primer.

The likely negative effect of directly incorporating RF into an SE adhesive primer on the degree of conversion and the mechanical characteristics of a polymerized resin may be another reason for decreased bond strength. This process negatively affects the mechanical properties of polymerized resins. For instance, it has been reported that the addition of 1% CHX to various resin mixtures with different hydrophilicities leads to a reduction in the modulus of elasticity (MOE) of polymerized resins by 27–48%. When directly incorporating RF into the primer composition, the possibility that it may react with the primer components and interfere with its optimal functioning cannot be ignored. However, further tests and chemical analyses are required in order to suggest this method with confidence.

In the present study, the bond strength of the RF/BL and RF-P/BL groups was similar. Despite the vastly different microscopic images, different potential mechanisms involved in reducing the bond strength in the 2 groups may explain this similarity. As discussed earlier, the obstruction of the path of resin penetration into the spaces between collagen fibers as a result of collagen cross-linking in the smear layer may be responsible, at least in part, for the reduction in bond strength in the RF/BL group; in the RF-P/BL group, however, the reduction in bond strength may be due to the long dura-
tion of dentin treatment with Clearfil SE Bond primer before applying the bonding agent, the low concentration of RF used or the likely negative impact of directly incorporating RF into the SE adhesive primer formulation.

Conclusions

Within the limitations of this study, cross-linking of collagen type I fibers did occur as the result of the 0.1 wt% aqueous solution of RF and the QTH light activation for 2 min. Pretreatment of the dentin surface with QTH light-activated RF decreased the μTBS of the Clearfil SE Bond as a SE adhesive system. No difference was detected between the application of 0.1 wt% RF alone to the dentin surface and the incorporation of 0.1% wt RF into the primer of the two-step SE adhesive system.

References


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